

NANOCRYSTALLINE HIGH ENTROPY MATERIALS: PROCESSING CHALLENGES AND PROPERTIES

Effect of niobium addition in FeCoNiCuNb_x high-entropy alloys

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In the design of high-entropy alloys (HEAs) with desired properties, identifying the effects of elements plays an important role. HEAs with eutectic microstructure can be obtained by judiciously modifying the alloy compositions. In this study, the effect of Nb addition to FeCoNiCuNb_x (x = 0.5, 5, 7.5, 11.6, 15) alloys was studied by varying the Nb concentration (at.%). FeCoNiCuNb_{0.5} HEA shows liquid phase separation to form Cu-rich and FeCoNiCu-rich phases. Detailed solidification paths are proposed for these alloys, which show eutectic, peritectic, and pseudo quasi-peritectic reactions. Increasing Nb content promotes the liquid phase separation tendency and causes the formation of Cu-rich spheres. The effect of Nb on the FeCoNiCu-rich phase was studied based on the nanoindentation and correlated with nanohardness. The compressive deformation properties of these alloys are studied at room temperature and high temperature and correlated with microstructure. Fractography results show the mode of fracture and are correlated with the microstructure obtained.

Introduction

Alloys with five or more principal elements are designated as high-entropy alloys (HEAs), complex concentrated alloys (CCAs), etc., which explore the central region of the phase diagram. HEAs with remarkable properties like high toughness, fracture strength [1, 2], high strength [3], good ductility, better cryogenic properties [4], good corrosion resistance [5], etc., are explored in the literature and hence can be considered as potential candidates for critical applications [6]. HEAs can also be designed to obtain conventional as well as unique microstructure, which is suitable for different applications [7, 8]. It is reported in open literature that HEAs show better strength and ductility [8]. The phase evolution and solidification pathways are also reported in HEAs, which show unique microstructures, like eutectic [3, 9, 10], quasi-peritectic [11], and phase separation [12, 13, 14]. There are different criteria established in the literature to predict the phases and microstructure formation. These criteria include the atomic size difference, valence electron concentration, electronegativity difference, enthalpy of mixing, and entropy of mixing [3, 15, 16]. It is to be noted that the usage of these parameters are not applicable to all HEAs, which make the design of HEAs cumbersome. The usage of thermodynamic simulation using CAPHAD approach is also reported in literature for the phase prediction in HEAs [17]. The study of phase evolution is important to understand the physical metallurgy involved in these alloys and to optimize the composition domain.

Recently, Huo and coworkers reported a eutectic HEA with excellent compressive strength and ductility [3]. The strength is attributed to the presence of Laves phase and a simple solid solution phase. FeCoNiCrNb_x eutectic systems reported in the literature show the presence of face-centered cubic (FCC) and Laves phase [9]. In these HEA systems, the amount of eutectic region increases with the Nb addition and reaches fully eutectic structure at 0.6 at.% of Nb. While increasing Nb further, the alloy shows formation of the eutectic region and Laves phase. The FeCoNiCrX (X = Zr, Ta, Mo) alloys with the combination of Laves and solid solution phases are reported in the literature [3, 18]. These studies suggest that the optimum combination of refractory elements alloying with the FeCoNiCr alloy can be designed to form multicomponent HEAs with better composite properties. Two-phase alloys with phase separation in the

as-cast condition are reported in the HEA systems [11]. The FeCoNiCuCr alloys show phase separation during undercooled conditions [19]. The eutectic FeCoNiCrNb_{0.25} shows the formation of γ'' precipitates during annealing at 750 °C [20]. It is also observed that eutectic system with the formation of precipitates will enhance the properties. Nanoindentation studies can be used for screening compositions for the development of HEAs [21]. Nanoindentation studies are reported for AlCoCrCuFeNi alloy and used to establish the deformation kinetics in HEAs [2].

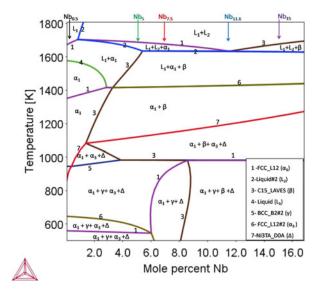


Figure 1: Phase diagram shows the equilibrium phase formation with Nb and the alloys selected for the study, where α_1 , FCC_L12; α_3 ,FCC_L12#2, Δ -NI3TA_D0A; L₁, Liquid; L₂, Liquid#2; β, C15_LAVES; and γ, BCC_B2#2.

The high-temperature flow behavior was reported in HEA systems. Recent reports show the development of HEAs with nanoprecipitates, which can replace the conventional Ni-based superalloys [7]. The room temperature and high-temperature mechanical properties reported for this system are comparable to Ni-based superalloys. The high-temperature tensile behavior in CrMnFeCoNi system suggests the strong dependence of the yield strength over the temperature [22]. The development of refractory HEAs makes the applicability of HEAs in hightemperature domain. The refractory HEAs with better ductility are recently developed by modifying the compositional domain and processing condition [23]. The effect of processing domains on the properties of HEAs shows the need to study the flow behavior at high temperatures [24]. The studies on eutectic alloys show that the microstructure of materials can be modified by varying the processing conditions [25].

In the present study, the effect of Nb addition on phase formation and solidification behavior in FeCoNiCu alloy was established. Nanoindentation study was carried out to establish the contribution of each phase in the selected alloy. Finally, the mechanical properties at room temperature and different temperatures of the studied HEAs are evaluated and correlated with microstructure.

Results and discussion

Thermodynamic prediction

Figure 1 shows the phase diagram predicted by thermodynamic calculations using ThermoCalc® software (Thermo-Calc Software, Solna, Sweden) with TCHEA2® database. The

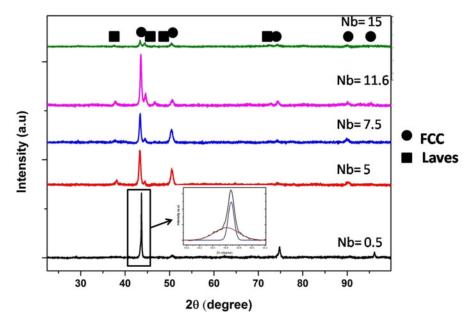


Figure 2: XRD pattern of FeCoNiCuNb_x alloys with varying concentration of Nb.



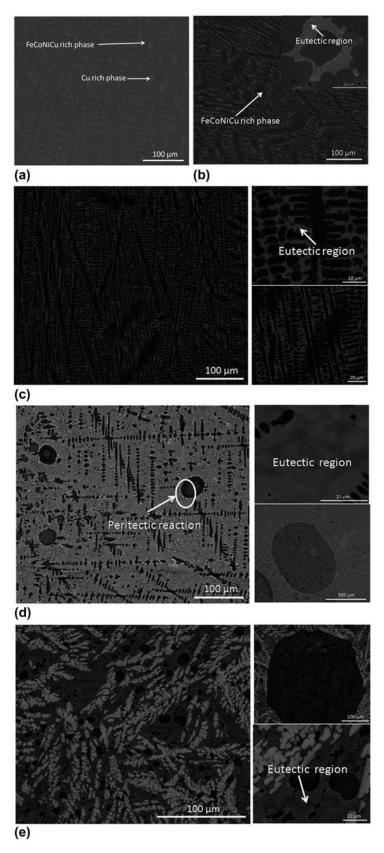


Figure 3: BSE images of suction casted FeCoNiCuNb_x alloys. (a) x = 0.5, (b) x = 5, (c) x = 7.5, (d) x = 11.6, and (e) x = 15.



variation in Nb in phase diagram shows eutectic point at 11.6 at.% of Nb. The alloys are chosen in such a way that eutectic region should be present in the microstructure. The eutectic high-entropy alloys (EHEAs) with simple solid solution phase and laves phase mixture exhibit a good combination of strength and ductility. According to the ThermoCalc® estimates, the primary phase is FCC phase below 11.6 at.% of Nb and Laves phase above 11.6 at.% of Nb. The thermodynamic estimates indicate that there is possibility of liquid phase separation. Using these calculations, the multicomponent FeCoNiCuNb_x (x = 0.5, 5, 7.5, 11.6, 15) HEAs are selected for the present study to understand the detailed microstructure formation during solidification.

Structural characterization

Figure 2 shows the X-Ray Diffraction (XRD) pattern of the studied HEAs with varying Nb content. The XRD pattern of 0.5 at.% of Nb in the HEAs shows only the presence of FCC phase. The deconvoluted first peak in the inset of Fig. 2 for 0.5 at.% of Nb condition shows the presence of two FCC phases with close lattice parameter. While increasing Nb content, it is found that the HEAs consist of FCC solid solution phase and Laves phase. The intensity of Laves phase was increased with increase in Nb, and it is approximately equal to FCC phase at 15 at.% of Nb, indicating the presence of equal volume fraction of each phases in the alloy.

Microstructural characterization

The microstructural characterization of the studied HEAs has been carried out using scanning electron microscopy (SEM) in backscattered electron (BSE) imaging mode to image distinctly the phases present in the microstructure using the elemental contrast. The phases in the microstructure are identified by compositional measurement using energy-dispersive spectroscopy (EDS). The representative micrographs are shown in Fig. 3, revealing the phase formation in these alloys with respect to the addition of Nb. It is clear that the solidification pathways are different with variation in Nb content. Figure 3(a) shows the SEM micrograph of FeCoNiCuNb_{0.5} alloy, showing the presence of only two phases, i.e., FeCoNi-rich phase (α_1) and Cu-rich (\alpha_2) phase. The XRD graph (Fig. 2) confirms that both phases are FCC structure with close lattice parameter. The composition of phases is given in Tables I, II, and III. It is clear that the primary FeCoNiCu-rich phase (α_1) is formed first from the liquid, followed by formation of Cu-rich phase (α_2) in the interdendritic region. The highly positive enthalpy of mixing for Cu with Fe and Co could explain the segregation of Cu-rich phase in the interdendritic region.

Figure 3(b) shows the SEM micrograph of 5 at.% of Nb content in the HEA. It is clear that the alloy consists of solid

TABLE I: Compositions of FeCoNiCu-rich phase.

Composition	Fe	Co	Ni	Cu	Nb
FeCoNiCuNb _{0.5}	28.41	27.05	25.13	19.23	0.18
FeCoNiCuNb ₅	27.35	26.17	24.51	20.37	1.60
FeCoNiCuNb _{7,5}	29.36	25.99	23.43	20.10	1.11
FeCoNiCuNb _{11.6}	29.93	25.33	24.06	18.34	2.34
FeCoNiCuNb ₁₅	27.04	21.73	25.38	20.36	5.48

TABLE II: Compositions of Laves phase.

Composition	Fe	Co	Ni	Cu	Nb
FeCoNiCuNb ₅	23.16	29.29	24.80	16.36	6.39
FeCoNiCuNb _{7.5}	25.44	28.00	25.13	16.24	5.19
FeCoNiCuNb _{11.6}	22.12	25.67	22.15	15.20	14.86
FeCoNiCuNb ₁₅	28.50	33.75	20.15	3.60	13.99

TABLE III: Composition of Cu-rich phase.

Composition	Fe	Со	Ni	Cu	Nb
FeCoNiCuNb _{0.5}	6.96	6.37	9.87	76.59	0.2
FeCoNiCuNb ₅	5.95	5.63	9.44	76.18	2.79
FeCoNiCuNb _{7,5}	4.40	4.17	5.71	83.52	2.20
FeCoNiCuNb _{11.6}	6.07	8.80	9.46	75.90	2.68
FeCoNiCuNb ₁₅	3.15	2.41	5.24	84.81	4.39

solution phases and eutectic region. The phases present are FeCoNiCu-rich phase (α₁), (Fe, Co)₂Nb-type Laves phase, and Cu-rich phase (α_2) . The eutectics are between Cu-rich phase (α_2) and (Fe, Co)₂Nb-type Laves phase. The solidification pathway can be explained as follows: (i) first primary (Fe, Co)2Nb-type Laves phase is formed from the liquid, followed by (ii) the formation of FeCoNiCu-rich phase (α_1) as secondary phase due to peritectic reaction, and (iii) finally, the remaining liquid undergoes eutectic reaction to form Cu-rich phase (α_2) and (Fe, Co)2Nb-type Laves phase. It is observed that Nb is mainly concentrated on the Laves phase. Figure 3(c) shows the SEM image of FeCoNiCuNb_{7.5} alloy. It is clear that the phase mixture remains same as that of FeCoNiCuNb₅ alloy. However, the volume fraction of eutectics is more and the interlamellar spacing is decreased with increase in the Nb content from 5 to 7.5 at.%. The solidification pathway for FeCoNiCuNb_{7.5} is also similar to that for FeCoNiCuNb5 alloy. Therefore, based on microstructural characterization, the following pseudo quasiperitectic reaction has been proposed for the alloys FeCoNi-CuNb₅ and FeCoNiCuNb_{7.5}.

L + FeCoNiCu-rich phase $(\alpha_1) \rightarrow$ Cu-rich phase (α_2) + (Fe, Co)₂Nb-type Laves phase.

Figure 3(d) shows the microstructure of FeCoNiCuNb_{11.6} alloy, revealing fine-scale eutectics. One may expect some liquid phase immiscibility in alloys that contain Cu together with Fe and Co. The presence of Nb could also accelerate the liquid phase separation tendency. In this alloy, large spheres of

Cu-rich phase are formed as the part of liquid phase separation. The spherical morphologies that appear inside the Cu-rich spheres indicate secondary liquid phase separation. This alloy also consists of Cu-rich phase (α_2) , FeCoNiCu-rich phase (α_1) , and Laves phase. The eutectic region was formed with the mixture of Cu-rich phase (α_2) and $(Fe, Co)_2Nb$ -type Laves phase. The solidification pathway can be explained as follows: (i) first, the Cu-rich phase (α_2) bubbles are formed from the liquid, followed by (ii) the formation of FeCoNiCu-rich phase (α_1) by means of peritectic reaction [i.e., L + Cu-rich phase

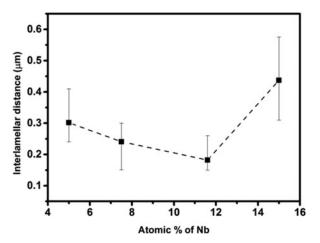


Figure 4: Interlamellar space variation of eutectic region in the FeCoNiCuNbx system.

 $(\alpha_2) \rightarrow \text{FeCoNiCu-rich phase } (\alpha_1)$, (iii) (Fe, Co)₂Nb-type Laves phase is precipitated from the remaining liquid, and (iv) finally, the liquid undergoes eutectic reaction between Cu-rich phase (α_2) and $(\text{Fe, Co})_2\text{Nb-type}$ Laves phase [i.e., L \rightarrow Cu-rich phase (α_2) + (Fe, Co)₂Nb-type Laves phase]. Liquid phase separation is reported in HEAs with Fe and Cu. The alloys with two phase mixture can undergo liquid phase separation based on the processing condition; this can be evident from the undercooling studies of FeCoNiCuCr alloy [19]. Phase separation was reported even in AlCoCrCuFeNi [26] alloys in nanoscale, which confirms that the high-entropy effect at high temperature is not able to avoid the phase separation. Therefore, based on microstructural characterization, the following pseudo quasi-peritectic reaction has been proposed for FeCo-NiCuNb_{11.6} HEA.

L + FeCoNiCu-rich phase $(\alpha_1) \rightarrow \text{Cu-rich phase } (\alpha_2) +$ (Fe, Co)₂Nb-type Laves phase.

Figure 3(e) reveals SEM micrograph of hypereutectic FeCoNiCuNb₁₅ HEA, indicating the presence of FeCoNiCurich phase (α₁), Cu-rich phase, and (Fe, Co)₂Nb-type Laves phase. The large Cu-rich spherical morphologies confirm the role of Nb in liquid phase separation in this system. The Nb will aggregate the liquid phase separation tendency. Here, the amount of Laves phase is more and it forms separately. The solidification pathway can be explained as follows: (i) first, the Cu-rich phase (\alpha_2) spheres are formed from the liquid,

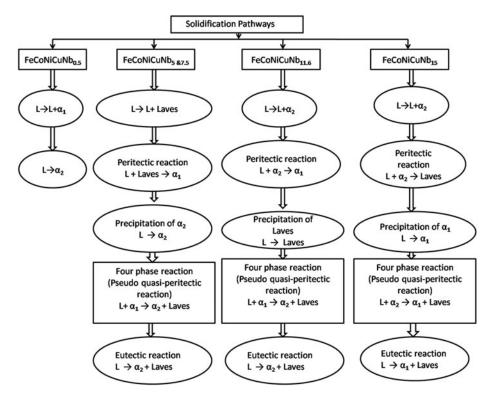


Figure 5: Detailed solidification pathways for FeCoNiCuNb_x system.



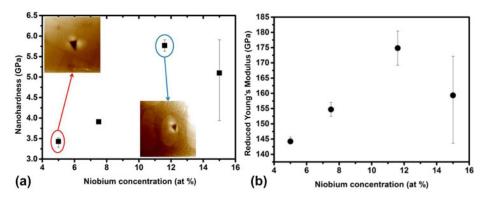


Figure 6: (a) Nanohardness variation of FeCoNiCu-rich phase and (b) reduced Young's modulus variation of FeCoNiCu-rich phase.

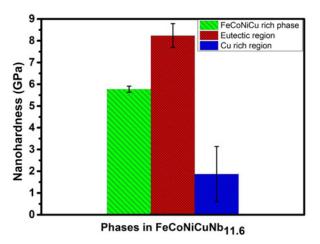


Figure 7: Individual phase hardness of FeCoNiCuNb_{11.6} HEA.

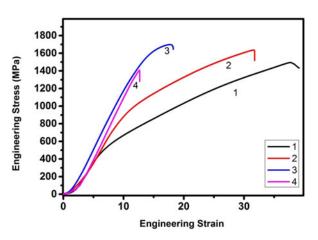


Figure 8: Room temperature engineering compressive stress–strain curve for (1) FeCoNiCuNb₅, (2) FeCoNiCuNb_{7.5}, (3) FeCoNiCuNb_{11.6}, and (4) FeCoNiCuNb₁₅.

followed by (ii) the formation of (Fe, Co)₂Nb-type Laves phase by means of peritectic reaction [i.e., L + Cu-rich phase (α_2) \rightarrow (Fe, Co)₂Nb-type Laves phase], (iii) FeCoNiCu-rich phase (α_1) is precipitated from the remaining liquid, and (iv) finally, the liquid undergoes eutectic reaction between FeCoNiCu-rich phase (α_1) and (Fe, Co)₂Nb-type Laves phase [i.e., $L \rightarrow$

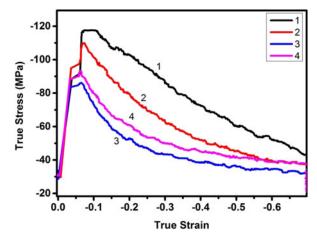


Figure 9: True stress–strain curve of the sample deformed at 1000 $^{\circ}$ C at a strain rate of 0.001 s $^{-1}$, where (1) FeCoNiCuNb₅, (2) FeCoNiCuNb_{7.5}, (3) FeCoNiCuNb_{11.6}, and (4) FeCoNiCuNb₁₅.

FeCoNiCu-rich phase (α_1) + (Fe, Co)₂Nb-type Laves phase]. Similarly, based on microstructural characterization, the pseudo quasi-peritectic reaction has been proposed for hypereutectic FeCoNiCuNb₁₅ HEA.

L + Cu-rich phase $(\alpha_2) \to FeCoNiCu$ -rich phase (α_1) + $(Fe, Co)_2Nb$ -type Laves phase.

From the SEM micrographs, it is clear that the phases are same from 5 to 15 at.% Nb, but the phase fractions obtained are varying based on the composition. From Fig. 4, it is clear that the interlamellar spacing decreases with Nb addition up to a certain value of Nb, then it will increase to hypereutectic region. From Table I, it is clear that the amount of Nb slightly increases in the FeCoNiCu-rich region and other elements almost maintain same ratio. It is found from Table II that the Nb in the Laves phase is increasing, which affects the strength of eutectic. Figure 5 shows the solidification pathways for these alloys.

Mechanical properties

Figure 6 shows the variation in nanohardness and reduced Young's modulus with respect to the Nb addition. The

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nanoindentation has been carried out in the FeCoNiCu-rich phase (α_1), which shows an increase in nanohardness and Young's modulus up to 11.6 at.% of Nb and then decreases at 15 at.% Nb. This may be attributed to the solid solution strengthening effect of Nb on FeCoNiCu-rich phase (α_1), and at 15 at.%, most of the Nb is partitioning in the Laves phase. The nanoindentation on the individual phases has been taken for FeCoNiCuNb_{11.6} alloy (Fig. 7). This alloy shows fine-scale eutectic microstructure less than 200 nm interlamellar spacing. The indentation results show that the eutectic region will

contribute to hardness more than the other phases. The FCC phase has hardness less than eutectic region, but it can provide adequate ductility. The Cu-rich region shows less hardness than other phases. The composite microstructure consisting of harder eutectic and softer dendritic phase will provide a good combination of strength and ductility in the material.

The room temperature compressive engineering stress-strain curve is shown in Fig. 8. The alloy with fine-scale eutectic (FeCoNiCuNb $_{11.6}$) shows high compressive strength (>1700 MPa) and ductility more than 17%. The presence of Cu-rich

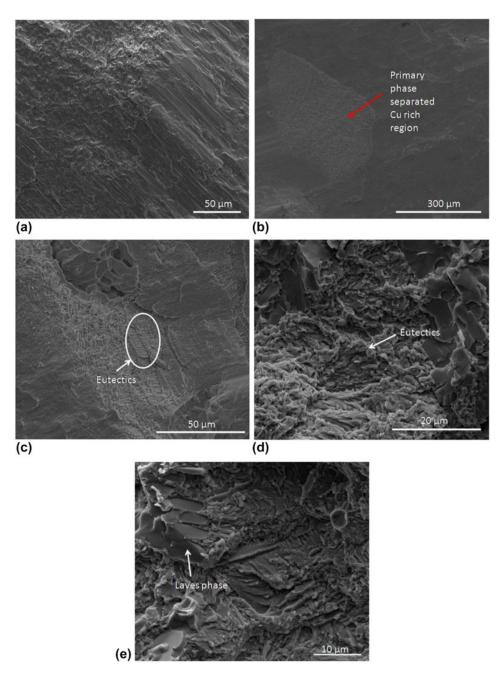


Figure 10: Fracture surface SEM images of compressed sample at room temperature: (a) FeCoNiCuNb_{7.5}, (b & c) FeCoNiCuNb_{11.6}, and (d & e) FeCoNiCuNb₁₅.



phase decreases the strength of the alloy compared with the reported eutectic alloys with FCC and Laves phase. From the compressive data, it is found that increase in Nb content causes increase in strength with a reasonable ductility up to 11.6 at. %. Further increase in Nb causes reduction in ductility as well as strength, which can be explained by phase formation and unique microstructural features during solidification. The hypereutectic alloy predominantly consists of Laves phase, which causes a detrimental effect in the compressive strength. The strength increase can be attributed to the decrease in the interlamellar spacing, which can be correlated with the Hall-Petch-type relationship between the yield strength and interlamellar spacing. In case of FeCoNiCuNb5 alloy, the sample did not break during the compression test but was sheared in a way that led to gradual reduction in strength after the maximum compressive strength. The other alloys experienced breakage after the maximum strength and were used for fracture studies.

Recently, processing maps have been developed for HEAs to determine their high-temperature deformation behavior [24, 27]. The processing maps for eutectic HEAs will make their applicability and help optimize the processing conditions. Understanding the high-temperature deformation behavior of the EHEAs will help identify the flow behavior and applicability of the alloy. The flow behavior of the sample at 1000 °C at a true strain rate of 0.001 s⁻¹ with a true strain of 0.7 is shown in the Fig. 9. The strength shown is comparable to that of the previously reported EHEAs. The flow characteristics are similar to those of eutectic alloys, where strength increases to a maximum value and then decreases to low flow stress condition. It is found that when Nb content increases, the flow stress decreases for a constant strain, which may be attributed to the softening of eutectic region. This can be confirmed from the flow characteristics of FeCoNiCuNb_{11.6}, where eutectic fraction was more than other phases. Fracture surface characteristics of FeCoNiCuNb_x (x = 7.5, 11.6, 15) are shown in Fig. 10. The fractography of FeCoNiCuNb_{7,5} alloy shows riverbed pattern, indicating brittle-type fracture. In case of FeCoNiCuNb_{11.6} and FeCoNiCuNb₁₅ alloys, the fracture surface consisting of eutectic region and the Cu-rich region shows the mixture of facetlike morphology and dimples in the microstructure. The hard and brittle Laves phase will act as a crack initiation site and reduces the ductility of the alloy, revealing brittle mode of fracture. The ductile Cu-rich region could act as a barrier for crack propagation. Thus, HEAs with 11.6 and 15 at.% Nb revealed mixed (i.e., both ductile and brittle) mode of fracture.

Conclusion

The effect of Nb addition on phase formation and solidification behavior in FeCoNiCu alloy has been studied. The variation in solidification pathway has been obtained in hyper- and hypoeutectic HEAs. Strong influence of Nb concentration on liquid phase separation is observed. The new pseudo quasiperitectic reactions, i.e., L + FeCoNiCu-rich phase $(\alpha_1) \rightarrow \text{Cu-}$ rich phase (α_2) + (Fe, Co)₂Nb-type Laves phase at 5, 7.5, 11.6 at.% Nb, and L + Cu-rich phase $(\alpha_2) \rightarrow FeCoNiCu$ -rich phase (α_1) + (Fe, Co)₂Nb-type Laves phase at 15 at.% Nb, have been proposed for the studied HEAs. The nanoindentation results show improvement in the hardness of FeCoNiCu-rich phase (α_1) , which is attributed to the solid solution strengthening of Nb. Increase in room temperature compressive strength is due to the decrease in interlamellar spacing of eutectic region. High-temperature deformation behavior shows the flow stress was similar to that of the reported EHEAs.

Experimental methods

All the alloys were prepared with elements of high purity (>99.9%) by using vacuum arc melting process. High vacuum was built in the melting chamber and purged with Argon gas for melting. The melting was carried out using the arc struck by a non-consumable tungsten electrode. The homogeneous samples were obtained by flipping the alloy button and remelting 6 times. To evaluate the mechanical properties, the arc-melted button was suction cast into cylindrical rod ($\phi =$ 6 mm diameter). The suction-cast sample was cut into the desired dimensions with an aspect ratio of 1.5:1. The hot deformation was carried out in Gleeble 3800® thermomechanical simulator (Dynamic Systems Inc, New York City, NY) at a deformation temperature of 1000 °C (1273 K) with constant strain rate of $0.001~\text{s}^{-1}$ at a true strain of 0.7. To avoid the barreling and frictional effects, the high-temperature lubricant (Ni paste and graphite foil) was used between the sample and anvil surface. The temperature of the sample was controlled and monitored by spot welding a K-type thermocouple on the surface of the sample. The in situ image of the indented phase was captured using Scanning Probe Microscopy (SPM) imaging. The room temperature test was carried out using Zwick machine with a strain rate of 0.001 s⁻¹. Nanoindentation test was carried out on the selected sample using Hysitron® machine (Bruker, Eden Prairie, Minnesota), and indentation was performed in different phases at the load of 5000 μN with a trapezoidal load function. The structural characterization was carried out using XRD (PANalytical X'pert pro® instrument; Malvern Panalytical Ltd., Royston, United Kingdom) with Cu K_{α} ($\lambda = 0.154056$ nm) radiation, operating at 45 kV and 30 mA, with step size of $2\theta = 0.017^{\circ}$. For microstructural characterization, the samples were polished using coarse and fine emery paper and the final polishing was carried out using alumina slurry. The polished samples were characterized using the BSE image mode in Quanta 400® SEM (ThermoFisher



Scientific, Hillsboro, Oregon) equipped with EDS detector (Bruker, Billerica, Massachusetts). Fractography was carried out on the room temperature compressed sample using SEM with scanning electron mode.

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